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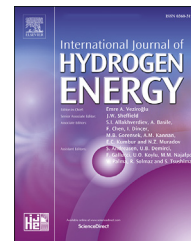
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The effect of prior austenite grain size on hydrogen embrittlement of Co-containing 18Ni 300 maraging steel

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ABSTRACT

The effects of the prior austenite grain sizes in hydrogen embrittlement of Co-containing 18Ni 300 maraging steel were studied employing Slow Strain Rate Testing (SSRT) in 0.6 M NaCl electrolyte under simultaneous cathodic polarization. The material was susceptible to hydrogen embrittlement in all investigated conditions. In addition, the examination of the fractured surface revealed that the presence of hydrogen in steel promotes the formation of quasi-cleavage regions and hydrogen-induced cracks along the grain boundaries. However, the refining of the prior austenite grain allowed an improvement in the HE resistance. Moreover, EBSD analysis showed that intergranular cracks propagated along to grain boundaries orientated to {001} planes parallel to normal direction, whereas they were deflected on {101} and {111} crystallographic planes.

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Introduction

The 18Ni maraging steels are a special class of materials, which present a combination of high-strength and high toughness. These mechanical properties have made 18Ni maraging steels widely used in critical applications, such as aeronautic and military industries [1]. These high-strength

steels are characterized by a low carbon content and are hardened by the precipitation reactions of fine intermetallic compounds during the aging treatment [2]. Typically, the combination of Co and Mo alloying elements leads to an effective hardening in 18Ni maraging steels [2,3]. In general, the heat treatment employed in alloys involves the solution annealing at elevated temperatures above 800 °C and then cooling to room temperature, which leads the formation of the

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well-known soft body martensite phase [1–3]. Finally, the material is submitted to an aging treatment at temperature ranging between 400 and 640 °C that promotes the precipitation of nano-sized intermetallic phases, such as Ni₃(Ti,Mo), Ni₃Ti, μ -phase and Fe₂Mo in the martensitic matrix, which are responsible for the excellent mechanical properties of these materials [1,4–8]. In addition, aging above 500 °C leads the reversion of austenite from the partially decomposition reaction of martensite by diffusion-controlled mechanism [6,9]. The presence of reverted austenite in 18Ni maraging steels has serious implications on their mechanical and magnetic properties [10–12].

It is known that the ultra-high strength 18Ni maraging steels can undergo environmentally assisted cracking, such as hydrogen embrittlement (HE) when exposed to hydrogen-rich environment [13–21]. In case of the Co-free 18Ni maraging steels several works considering the topic can be found in the literature [13–18]. Rao et al. [13], studied the effect of stress concentration factor in 18Ni 2400 (T-350) maraging steel aged at 510 °C for 1 h in synthetic seawater environment. These authors observed a drastically reduction in notched tensile strength and time to fracture of steel after testing, as well as, intercrystalline fracture mode on crack region. In addition, hydrogen assisted cracking of T-250 maraging steel aged at 480 °C for 2 h has been reported in air when the relative humidity (RH) was $\geq 30\%$ [15]. Moreover, Tsay et al. [17], demonstrated that T-200 maraging steels aged at 427 °C, 482 °C and 538 °C were susceptible to HE in the saturated H₂S solution. On the other hand, considering the Co-containing 18Ni maraging steel a significant amount of studies dedicated to understand this phenomenon are also found [19–24]. In the 1970s Wei and co-workers [19–21] revealed that different grades of 18Ni maraging steels exhibited similar crack growth kinetics under gaseous hydrogen atmosphere. Reddy et al. [22], reported that tensile properties and fracture characteristics of 18Ni 250 maraging steel have been correlated with the quantity of hydrogen picked up by the steel, additionally, they observed a change in fracture surface from ductile dimples to mixed mode, intergranular separation and transgranular cleavage as the amount of absorbed hydrogen increased. However, few works are devoted to investigate HE in grade 300 18Ni maraging steel. Santos et al. [23], studied HE of ultra-high strength 18Ni 300 maraging steel and observed that the intergranular cracks propagate mainly along grain boundaries with {001}//ND fiber orientation. Besides, Béres and collaborators [24] employing *in situ* synchrotron X-ray diffraction in this material to revealed that the {001} parallel to normal direction (ND) are crystallographic planes on which undergo highest magnitude of lattice strain during tensile loading, further this study showed that the cracks were arrested when faced {111}//ND.

Currently, several studies have been conducted to investigate the role of microstructure into the HE of ultra-high strength steels [25–31], such as austenitic stainless steel [32], dual-phase (DP), quenched and partitioned (Q&P), and twinning induced plasticity (TWIP) steels [33], 17-4 PH stainless [34], low-alloy steels [35] and 18Ni maraging steels [14,16,17,36]. Wang et al. [36], claimed that in over-aged condition Co-containing TM210 maraging steel exhibited higher resistance to HE, additionally, they also observed that HE

susceptibility depends on reverted austenite content. In addition, it has been already reported that reverted austenite acts as irreversible hydrogen traps in maraging steels, which prevents the transport of hydrogen inward to the highly strained region [16,17]. Furthermore, Tsay et al. [14], demonstrated that in the coarse-grained structure would raise the HE susceptibility of T-200 maraging steel and also the fracture features.

Therefore, the aim of this work was to evaluate the influence of grain size in the HE of the 18Ni 300 maraging steel. In order to obtain samples with different grain size the material was solution annealed at four temperatures combined with two aging conditions. In addition slow strain rate tests (SSRT) were carried out in air and under hydrogen environment. Finally, electron backscattering diffraction (EBSD) was performed on fractured surface.

Material and experimental

Material

The material used in this research was the commercial Co-containing 18Ni 300 maraging steel. The chemical composition of this alloy in weight percent is 18.28 Ni, 9.41 Co, 4.73 Mo, 0.73 Ti, balance Fe.

Heat treatments and XRD characterization

In order to obtain samples with different grain sizes, the solution annealing treatment of the 18Ni 300 maraging steel (1.5 cm × 1.0 cm × 1.5 mm) was performed at 840 °C, 950 °C, 1050 °C and 1150 °C in a muffle furnace during a period of 1 h followed by an air cooled to room temperature and then divided in two groups of samples. The first group of samples was aged at 480 °C for 3 h and the second one at 560 °C for 1 h. After aging, the samples were air cooled.

X-ray diffraction (XRD) measurements were also carried out on the samples aged at 560 °C for 1 h in order to detect any reverted austenite. The samples were sanded up to 400 mesh (SiC paper). The measurements were carried out using a Philips® X'Pert Pro diffractometer with a radiation of CoK α (0.1789 nm). The 2 θ angle ranged from 45° to 104°. This measurement was carried out only for the second group of samples since aging the samples at 480 °C up to 50 h does not form reverted austenite as reported in previous study [23].

Prior austenite grain size and microstructure analysis

The samples used for the determination of the prior austenite grain size were sanded using SiC paper up to 2500 mesh, washed in distilled water and blow dried followed by polishing using alumina (Al₂O₃) of 1 and 0.05 μ m to achieve optical quality of the images. Prior austenite grain size was revealed using an electrolytic etching of 10 V with stepped time. The electrolyte used was 20% chromic acid (H₂CrO₄) solution. The images were taken by Optical Microscopy (MO) using a Zeiss Microscope model Axio Imager M2m. Prior austenite grain size was computed using the software ImageJ following the recommendations of the ASTM E112-96 [37]. Martensitic structure

of 18Ni 300 maraging steel was also revealed using a chemical etching with HNO₃ (4%) in methanol by immersing the samples in this reagent up to 10 s by Philips® XL–30 Scanning Electron Microscopy (SEM).

Slow strain rate test (SSRT)

Cylindrical samples manufactured following the ASTM E8/E8M-11 standard [38] were used for SSRT. The produced samples were heat treated as mentioned above (see Section Heat treatments and XRD characterization) in a vacuum sealed quartz tube, air cooled and then categorized as shown in Table 1.

The SSRT was performed in samples in air and in an aqueous solution of 0.6 M NaCl under a simultaneous application of a cathodic potential of $-1.2 V_{SCE}$ [39]. The strain rate of $1.0 \times 10^{-6} s^{-1}$ was applied for samples tested in air. On the other hand, for the samples tested in an aqueous solution the

strain rate used was $1.0 \times 10^{-5} s^{-1}$ [39]. Prior to each SSRT test all the samples were sanded using SiC paper up to 600 mesh, washed in distilled water and blow dried. A potentiostat/galvanostat (Autolab, Metrohm-EcoChemie) was used to supply the cathodic potential during the SSRT in solution. A three-electrode electrochemical cell was used where the reference electrode was a saturated calomel electrode (SCE) and the counter electrode was a platinum sheet. The working electrode was the samples. Before the SSRT, the samples were immersed in the solution for 30 min in order to determine the open circuit potential (OCP). After reaching the OCP ($-0.42 V_{SCE}$), the samples underwent cathodic charging for 24 h prior to the SSRT. The tests were performed in a Cortest® machine model Constant's Extension Rate Test with a load cell of 44 kN. All the fractured surfaces after SSRT were observed by SEM. Electron Backscatter Diffraction (EBSD) was also used to analyze the preferred path of the cracks regarding the crystal orientation in the region of crack tip. For this analysis, the fractured region of the samples was prepared by sanding using SiC up to 600 mesh followed by polishing with 6, 3, 1 μm diamond paste and subsequently a final polishing of 0.5 μm colloidal silica suspension. The SSRT were reproduced in duplicate.

Table 1 – Categorized of samples tested on SSRT.

Group	Samples	Annealing	Aging
Group A	S1-480	840 °C/1 h	480 °C/3 h
	S2-480	950 °C/1 h	
	S3-480	1050 °C/1 h	
	S4-480	1150 °C/1 h	
Group B	S1-560	840 °C/1 h	560 °C/1 h
	S2-560	950 °C/1 h	
	S3-560	1050 °C/1 h	
	S4-560	1150 °C/1 h	

Results and discussion

Prior austenite grain size growth

Optical micrographs of 18Ni 300 maraging steel from samples of group A are shown in Fig. 1. It is noticed that the prior

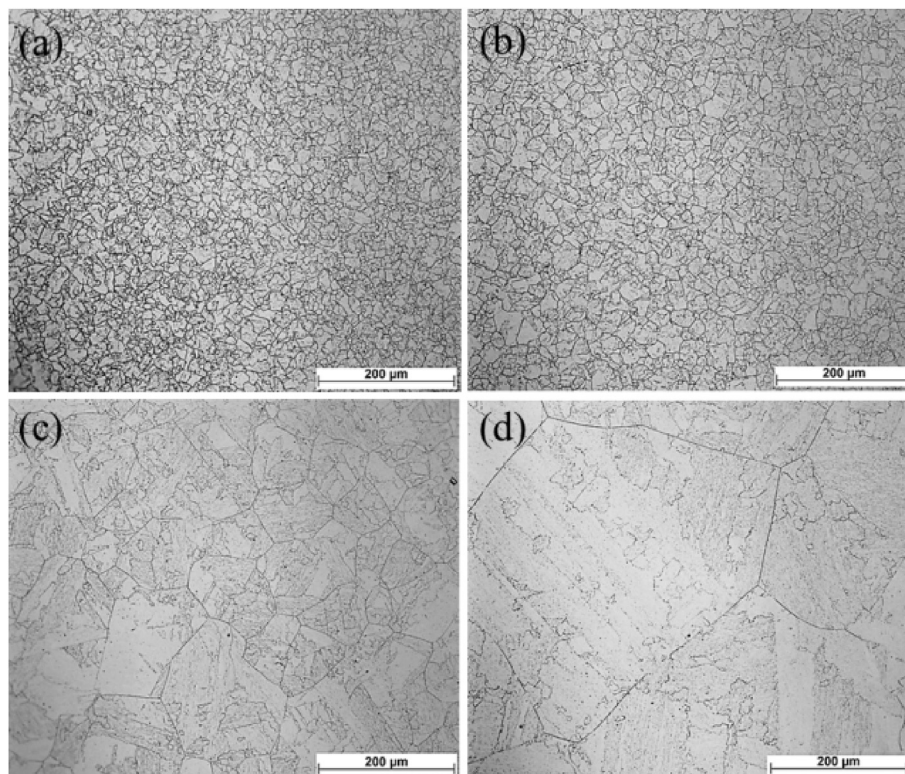


Fig. 1 – Optical micrograph of Co-containing 18Ni 300 maraging steel samples (a) S1-480, (b) S2-480, (c) S3-480 and (d) S4-480.

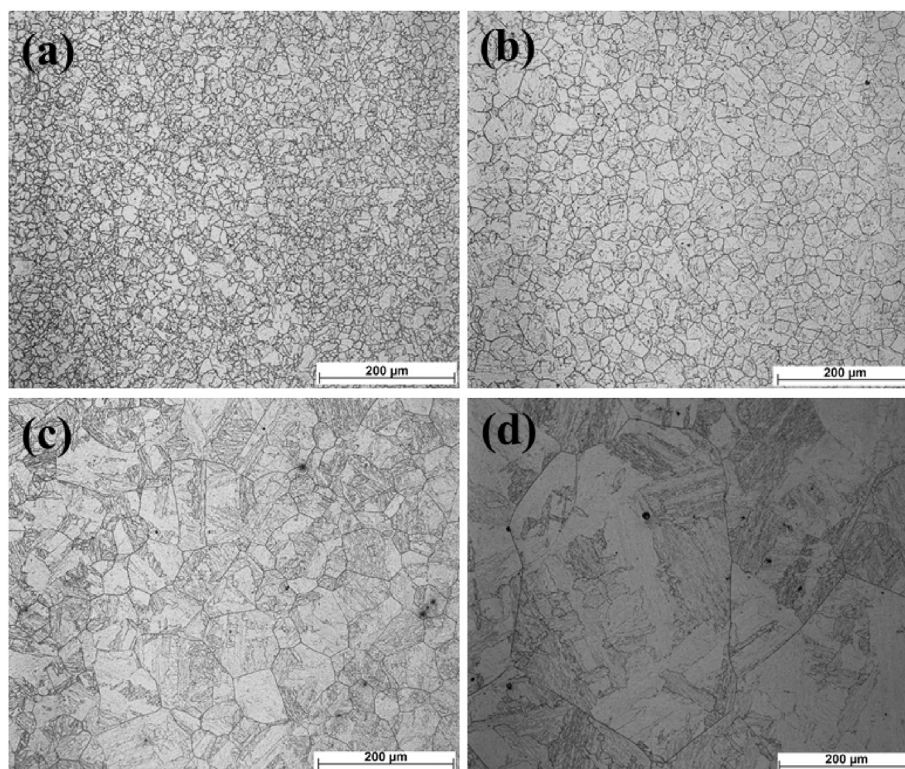


Fig. 2 – Optical micrograph of Co-containing 18Ni 300 maraging steel samples (a) S1-560, (b) S2-560, (c) S3-560 and (d) S4-560.

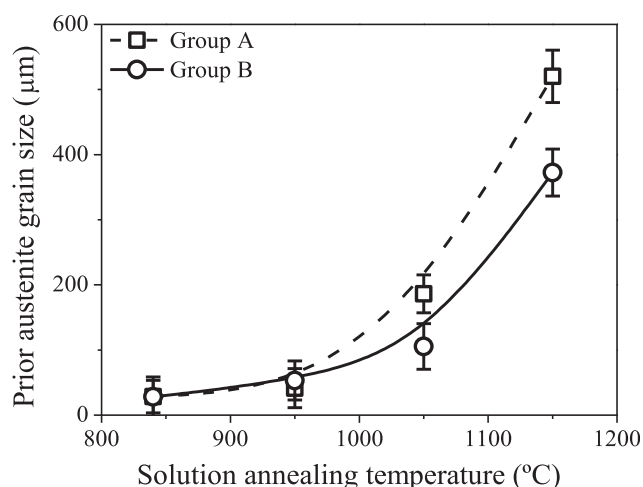


Fig. 3 – Prior austenite grain size of Co-containing 18Ni 300 maraging steel as function of solution annealing temperature.

austenite grain sizes quiet close for samples S1-480 and S2-480 around 28 and 48 μm , as can be seen in Fig. 1a and b, respectively. It is known that the grain growth results from grain boundary motion [40], thus, in this condition precipitates that act as barriers preventing the growth of the grain may exist. In addition, in the samples S3-480 and S4-480 was observed a meaningful increase of grain sizes changing to 186

and 520 μm as seen in the optical micrographs displayed in Fig. 1c and d, respectively. Similar results were observed for samples of group B, as seen in Fig. 2. Therefore, it is suggested that solution annealing treatment at temperatures higher than 1000 $^{\circ}\text{C}$ these barriers cease to exist by dissolving the precipitates in the matrix, leaving the dislocations free to move and thus allowing an increase of the prior austenite grains. Grain size average values for both groups as a function of solution annealing temperature is shown in Fig. 3. It is clearly seen that prior austenite grain size followed an exponential tendency growth. Furthermore, in Fig. 4 SEM examination revealed that the microstructure of 18Ni 300 maraging steel in all heat treated conditions were very similar and characteristic of these materials. Fig. 4a and b shows SEM micrographs of samples S4-480 and S4-560, respectively. One can see that the martensitic matrix consisted of well-defined martensite laths.

The quantitative measure of the prior austenite grain size was fixed according to ASTM E 112-96 [37], which ranks the grain on a scale ranging from 00 to 14.0. This classification is called grain size number. Table 2 shows this classification for the average grain size obtained in this research. The maximum grain size corresponds to 00 on this scale. This means that for the solution annealing at 1150 $^{\circ}\text{C}$ for 1 h for both aging conditions, the grain achieved the maximum ASTM value. The heat treatment of aging after the solution annealing did not influence the grain size. These results are in good agreement with results reported by Lima Filho et al., [43].

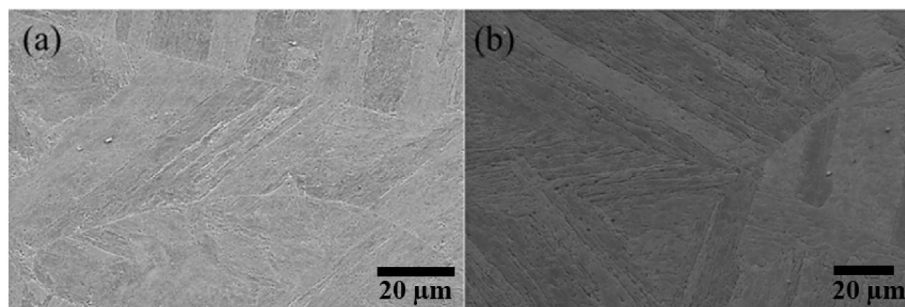


Fig. 4 – SEM micrograph of Co-containing 18Ni 300 maraging steel samples (a) S4-480 and (b) S4-560.

Table 2 – Grain size number classification for the samples solution annealed at several temperatures and aged at 480 °C for 3 h.

ASTM Grain size number	Temperature (°C)	Mean Diameter (μm)
7.5	840	28.6
6.0	950	41.3
2.0	1050	186.2
00	1150	520.2

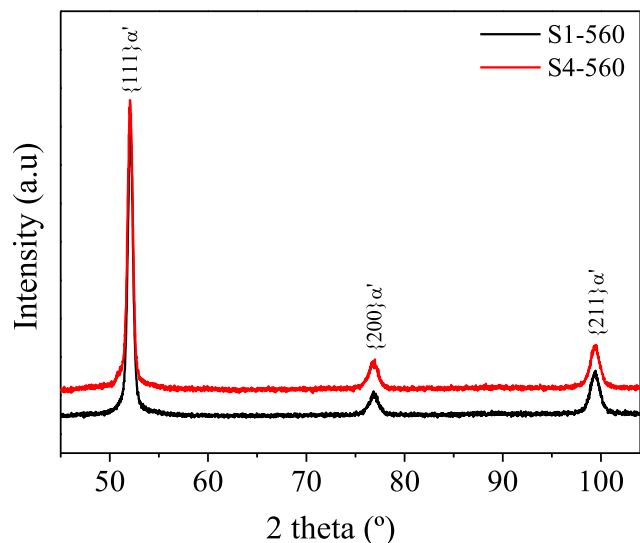


Fig. 5 – XRD pattern of Co-containing 18Ni 300 maraging steel samples (a) S1-560 and (b) S4-560.

XRD analysis

In order to detect the presence of reverted austenite after aging treatment at 560 °C XRD measurement was performed. Fig. 5 shows the XRD patterns for samples S1-560 and S4-560. Only peaks from BBC structure can be seen on this diffractogram corresponding to the martensite phase (α'). It is well-established in literature that aging of 18Ni 300 maraging steel at temperatures above 500 °C for long periods of time can promote formation of reverted austenite as a consequence of the partial dissolution of $\text{Ni}_3(\text{Ti}, \text{Mo})$ [6–10]. However, in this

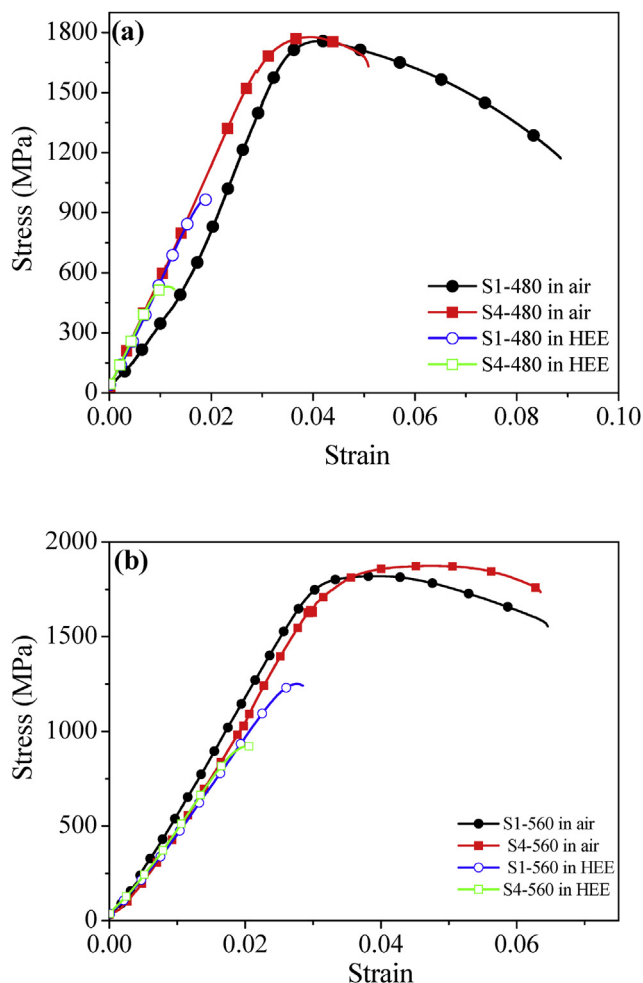


Fig. 6 – (a) SSRT stress–strain curves for samples S1-480 (28.6 μm) and S4-480 (520.2 μm) in air and under dynamic hydrogen charging in 0.6 M NaCl solution and (b) SSRT stress–strain curves for samples S1-560 (28.8 μm) and S4-560 (372.4 μm) in air and under dynamic hydrogen charging in 0.6 M NaCl solution.

research, no diffraction peaks corresponding to reverted austenite nor the peaks of the intermetallic compounds were found or the volume fractions of these phases are below the detection limit of technique that is around 5% [41]. Moreover, Sha et al. [6], reported that after aging the 18Ni 300 maraging

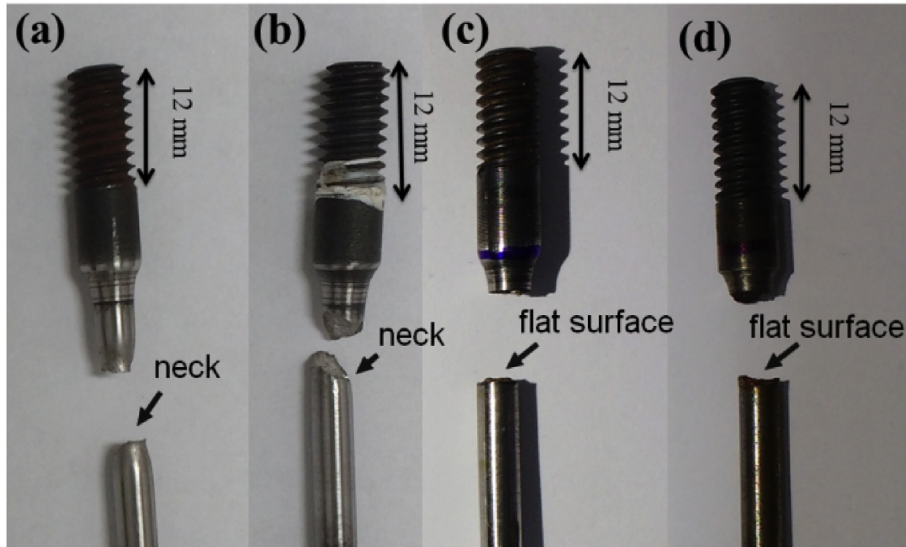


Fig. 7 – Photographic of lateral-view of fractured samples S1-480 and S4-480 tested in air and in HEE.

Table 3 – Mechanical properties taken from SRRT for the samples of 18Ni 300 maraging steel tested in air and in HE.

Sample	UTS in air, MPa	UTS in HEE, MPa	Strain in air, %	Strain in HEE, %	EI(Strain), %	EI(UTS), %
S1-480	1764	965	8.9	1.9	78.6	45.3
S4-480	1786	527	5.1	1.3	74.5	70.5
S1-560	1825	1243	6.4	2.8	56.2	31.2
S4-560	1873	929	6.3	2.0	68.2	50.4

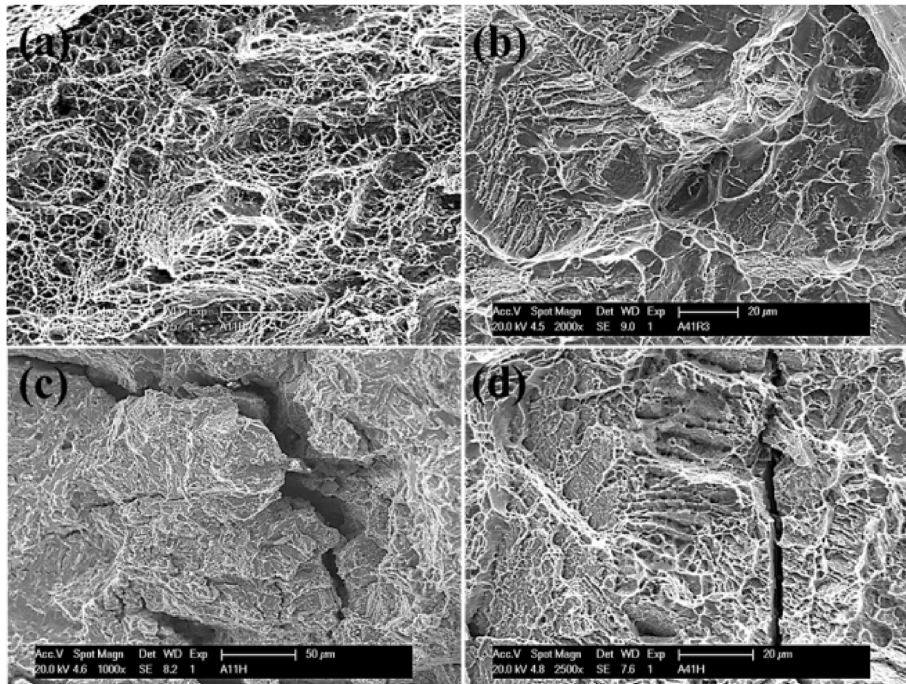


Fig. 8 – SEM fractographs of samples (a) S1-480 in air, (b) S4-480 in air, (c) S1-480 in HEE and (d) S4-480 in HEE.

steel at 510 °C for 128 h the volume fraction of precipitates were less than 4 wt percent. Besides, Pardal et al., observed that the amount of reverted austenite in 18Ni maraging steel

aged at 560 °C for 1 h is around 5%. Therefore, reverted austenite may exist in samples S1-560 but it is too low to be detected by XRD.

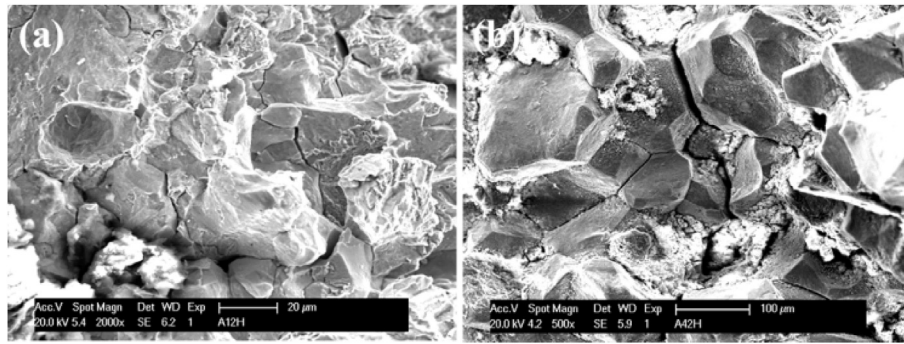


Fig. 9 – SEM fractographs of samples (a) S1-560 and (b) S4-560 in HEE.

Hydrogen embrittlement susceptibility by SSRT

In this section, the results of SSRT are presented for both samples of the 18Ni 300 maraging steel tested in air (inert medium) and tested in aqueous solution of 0.6 M NaCl under simultaneous cathodic polarization of $-1.2 V_{ECS}$, herein called as hydrogen environment embrittlement (HEE).

Fig. 6a shows the SSRT stress-strain curves of samples S1-480 and S4-480 tested in air and in HEE. These conditions of solution annealing were chosen taking into account the smallest and largest grain sizes found in this research. The grain sizes number of steel at the corresponding condition can be seen in brackets. The samples tested in air presented a similar mechanical behavior, thereby, ultimate tensile strength (UTS) for sample S1-480 was around 1764 MPa, while for sample S4-480 an average value of 1786 MPa, however, the sample S1-480 presented a higher elongation than the sample S4-480. On the other hand, the samples tested in HEE presented a drastic reduction of mechanical properties in comparison with the ones tested in air, Fig. 6a. Thus, the samples S1-480 and S4-480 underwent a reduction to 965 and 527 MPa, respectively in their UTS values. Additionally, the steel suffered little or almost no plastic deformation prevailing the fragile fracture. Fig. 7 shows a photographic of lateral-view of fractured samples S1-480 and S4-480 tested in air and in HEE. Fig. 7a–b, tested in air, indicate that the samples exhibited extensive plastic deformation with a neck formation surrounded by a shear lip which is typical of ductile fracture. In contrast, in the samples tested in HEE (Fig. 7c–d) a flat surface without necking was observed confirming the brittle fracture behavior, as seen in Fig. 7a–b. The SSRT stress-strain curves of samples S1-560 and S4-560 tested in air and in HEE are shown in Fig. 6b. These samples exhibited similar mechanical behavior of those found in samples S1-480 and S4-480, Fig. 6a. For tests conducted in air, the samples S1-560 and S4-560 showed a ductile performance with the UTS of 1825 and 1873 MPa, respectively, whereas, for tests performed in HEE, the UTS of these samples presented a reduction to 1243 and 929 MPa. In order to evaluate the HE damage, an embrittlement index (EI) taking into account of the mechanical properties of material was introduced by the following expression [36]:

$$EI(X) = \frac{X_{air} - X_{HE}}{X_{air}} \quad (1)$$

where X is the property under investigation, in this case, the relative UTS and strain reduction. The increase of this index indicates that the mechanical property is more sensible to hydrogen damage. The mechanical properties of S1-480, S4-480, S1-560 and S4-560 obtained from SSRT stress-strain

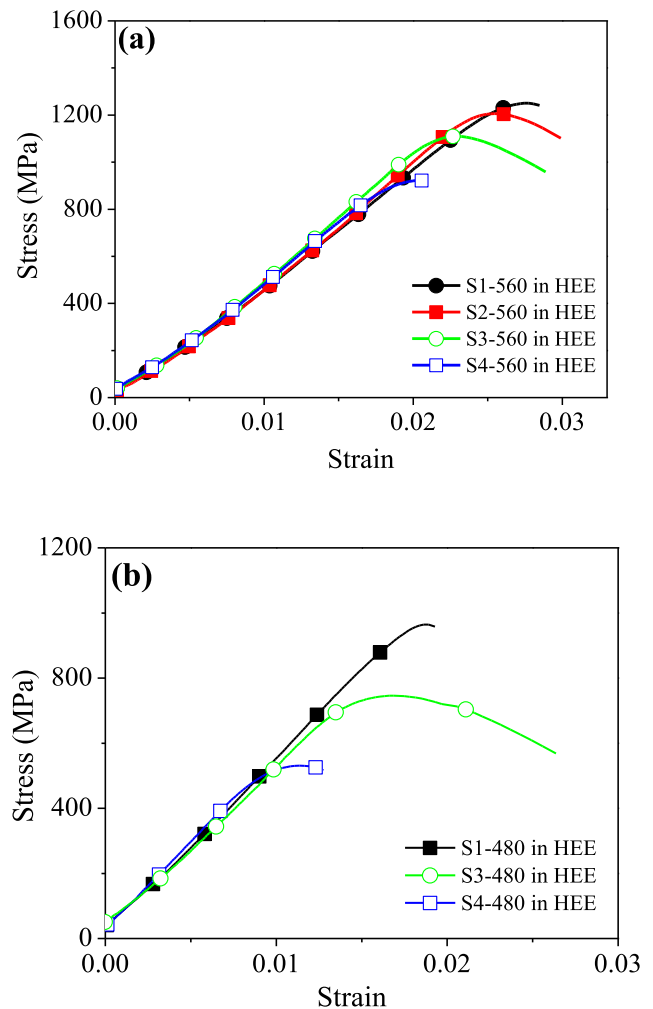


Fig. 10 – SSRT stress–strain curves for samples (a) S1-560 (28.8 μm), S2-560 (53.1 μm), S3-560 (105.3 μm) and S4-560 (372.4 μm) (b) S1-480 (28.6 μm), S3-480 (186.2 μm) and S4-480 (520.2 μm) under dynamic hydrogen charging in 0.6 M NaCl solution.

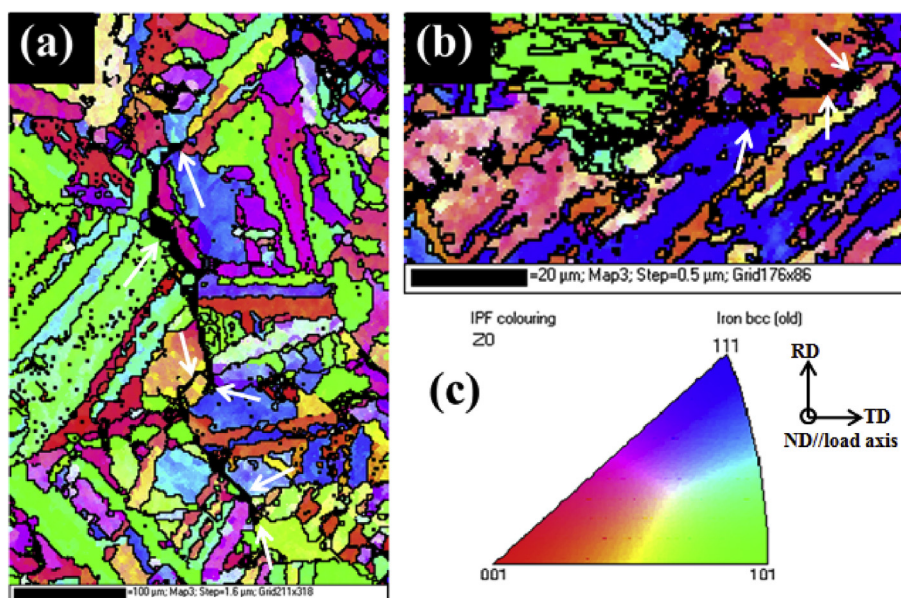


Fig. 11 – EBSD orientation map from the sample (a) S4-560, (b) S1-480 and (c) corresponding standard IPF color triangle.

curves (Fig. 6a–b) are shown in Table 3, as well as, their *EI*. This index indicates that the samples solution annealed at higher temperature are more susceptible to HE in both aging conditions. In addition, the samples aged at 560 °C for 1 h were less susceptible to HE than the ones aged at 480 °C for 3 h. Thereby, suggesting the presence of reverted austenite and/or other precipitates in these samples at this condition. It was reported that maraging steel containing reverted austenite exhibited the best resistance to HE [23,36].

Furthermore, Fig. 8 shows a SEM view of the fractured surfaces of the samples obtained from SRRT stress-strain curves shown in Fig. 6. In Fig. 8a–b is possible to see the fractograph of samples S1-480 and S4-480, respectively, tested in air. These figures present a great amount of dimples which characterized an extensive plastic deformation before the fracture of steel. Moreover, similar fractographs were obtained from samples S1-560 and S4-560 tested in air and the results were also typical of ductile fracture. SEM micrographs of S1-480 and S4-480 tested in HEE revealed a few dimples, quasi-cleavage regions with the presence of hydrogen-induced cracks propagated into fractured surface, indicating that the material suffered a minimum of plastic deformation before fracture, as can be observed in Fig. 8c–d. Moreover, SEM of fractured surface of samples S1-560 and S4-560 tested in HEE exhibited quasi-cleavage regions characteristic of brittle fracture with hydrogen-assisted intergranular cracks, as seen in Fig. 9a–b. Wang et al. [36], demonstrated that hydrogen-induced cracks propagating along the grain boundaries are a resulting of hydrogen concentration in this regions.

SSRT stress-strain curves for samples S1-560, S2-560, S3-560, and S4-560 are shown in Fig. 10a, while the curves for samples S1-480, S3-480 and S4-480 in Fig. 10b. There is no considerable change in the elastic regime regardless the solution annealing temperature in both aging conditions. Furthermore, one can see that the UTS decreases as the prior

austenite grain size of steel increases. It suggested that there may be a relationship between the prior austenite grain size and HE susceptibility. According to Park et al. [42], fine-grained showed superior HE resistance when compared with coarse-grained in API 2 W grade 60 steel. Additionally, Takasawa et al. [35], studied HE of high-strength low-alloy steels and observed that the grain refinement and a reduction in dislocation density are effective in reducing the susceptibility to embrittlement of material.

EBSD maps on the region around the crack propagating on the fracture surface of sample S4-560 and S1-480 are showed in Fig. 11a and b, respectively. The image refers to the crystal orientation in the region of the cracks tips. In this kind of EBSD examination (Fig. 11a–b) the cracks are seen in the map as thicker black lines corresponding the region where no indexing were possible. Additionally, due to the great deformation induced by the fracture of samples, same regions around the crack path were difficult to record a good indexing pattern, thus same black points can be seen as inside the grains. As can be seen in Fig. 11a–b, the cracks initially propagated along red grains, which are orientated through {001} planes parallel to normal direction (//ND), regardless the aging condition. It's known that cleavage fracture occurs preferentially through these crystallographic planes in bcc crystals [43]. Moreover, EBSD also revealed that the crack continued to propagate intergranularly avoiding the grain {101} and {111}//ND, seen in green and blue color in Fig. 11a–b and therefore, passing along their grain boundaries, as indicated by white arrows. These orientations have high crack propagation resistance due to their higher planar density. Previous worked reported that in 18Ni (300) maraging steel the {001} planes can accumulate approximately four times larger strains than {111} planes [24]. Thereby, it can be concluded that the grain orientated to {001}//ND were more prone to hydrogen induced cracks to propagate, due this family's lower planar atomic density, meaning more free sites for the crack propagation, on the

other hand, the denser crystallographic planes {101} and {111} showed higher resistance to hydrogen assisted cracks. These results are in good agreement with other authors [23,36].

Conclusions

The effect of prior austenite grain size of Co-containing 18Ni 300 maraging steel was investigated. Based on the achieved results, the following conclusions can be drawn:

1. The prior austenite grain size was a function of the solution annealing temperature and regardless of the aging temperature.
2. Co-containing 18Ni 300 maraging steel was susceptible to HE at all heat treatment conditions studied. However, samples with smallest prior austenite grains sizes were less susceptible to HE. The prior austenite grain refining improves the HE resistance of material.
3. The cracks for the samples tested in HE propagated along the grain boundaries causing great damage on the fractured surface. The paths for intergranular hydrogen induced crack propagation were in grain orientated along to {001}//ND, while the other orientations {110} and {111}//ND were more resistant to crack propagation.

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