

Preliminary *in vitro* evaluation of the anti-proliferative activity of guanyldrazone derivatives

PAULO H. B. FRANÇA²
EDEILDO F. DA SILVA-JÚNIOR²
PEDRO G. V. AQUINO¹
ANTÔNIO E. G. SANTANA¹
JAMYLLÉ N. S. FERRO³
EMILIANO DE OLIVEIRA BARRETO³
CLÁUDIA DO Ó PESSOA⁴
ASSUERO SILVA MEIRA⁴
THIAGO M. DE AQUINO²
MAGNA S. ALEXANDRE-MOREIRA⁵
MARTINE SCHMITT⁶
JOÃO X. DE ARAÚJO-JÚNIOR^{2*}

¹ *Natural Resources Research Laboratory
Institute of Chemistry and Biotechnology
Federal University of Alagoas, Maceio, Brazil*

² *Laboratory of Medicinal Chemistry
School of Nursing and Pharmacy
Federal University of Alagoas, Maceio, Brazil*

³ *Cell Biology Laboratory, Biological and
Health Sciences, Federal University of
Alagoas, Maceio, Brazil*

⁴ *Experimental Oncology Laboratory
Department of Physiology and Pharmacology
Federal University of Ceara, Fortaleza, Brazil*

⁵ *Institute of Biological and Health Sciences
Laboratory of Pharmacology and Immunity
Federal University of Alagoas, Maceio, Brazil*

⁶ *Laboratoire d'innovation Thérapeutique
Faculté de Pharmacie, Université de Strasbourg
Illkirck, France*

Guanyldrazones have shown promising antitumor activity in preclinical tumor models in several studies. In this study, we aimed at evaluating the cytotoxic effect of a series of synthetic guanyldrazones. Different human tumor cell lines, by including HCT-8 (colon carcinoma), MDA-MB-435 (melanoma) and SF-295 (glioblastoma) were continuously exposed to guanyldrazone derivatives for 72 hours and growth inhibition of tumor cell lines and macrophages J774 was measured using tetrazolium salt (MTT) assay. Compounds **7**, **11**, **16** and **17** showed strong cytotoxic activity with IC_{50} values lower than $10 \mu\text{mol L}^{-1}$ against four tumor cell lines. Among them, **7** was less toxic to non-tumor cells. Finally, obtained data suggest that guanyldrazones may be regarded as potential lead compounds for the design of novel anticancer agents.

Keywords: guanyldrazone, cytotoxic effect, synthesis, MTT assay

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* Correspondence; e-mail: jotaaranjo2004@gmail.com

Guanylhydrazones represent a class of compounds showing interesting pharmacological activities on several levels; they derive from the same aminoguanidine chemotype with mixed hydrogen bond acceptor and donor properties as well as being able to establish electrostatic interactions (1, 2). Moreover, they already led to various active derivatives, as illustrated in Fig. 1 (3). Guanidine derivatives have been reported to be efficient anti-cancer agents. Previous reports showed that *m*-iodobenzylguanidines (MIBG), pyridylcyanoguanidine (CHS 828) and mitoguazone (MGBG) have shown promising antitumor activity in preclinical tumor models, while mitoguazone showed useful clinical activity for the treatment of malignant lymphoma, carcinoma of the head, neck and esophageal and non-small cell lung cancer (4). More recently, a series of imidazo[2,1-*b*]thiazoleguanylhydrazones was reported for their antitumor activity and was able to induce apoptosis in HT29 and HL-60 cell lines (5). However, no structure-activity relationship of imidazothiazole compounds was available. Sixteen guanylhydrazone coactivator binding inhibitors for estrogen receptor showed good to moderate activity, but no cytotoxicity data were shown for these compounds (6).

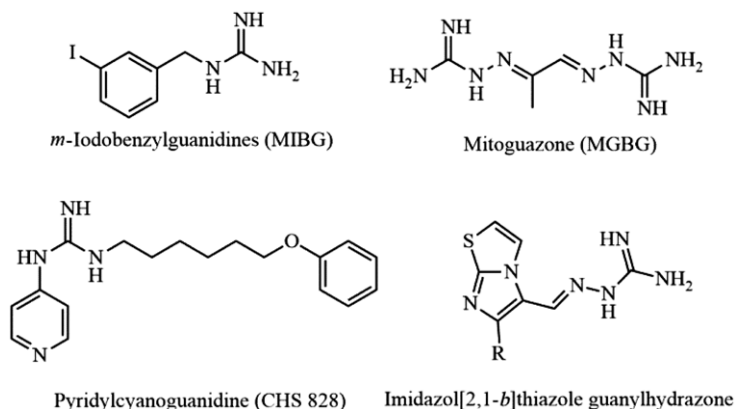


Fig. 1. Cytotoxic guanylhydrazones previously reported in literature (4).

After identification of the guanidine moiety, as the key pharmacophoric group for activity, we decided to replace the imidazothiazole ring by a phenyl ring, which offered the advantage of scaffold modification. In particular, our objective was to perform a systematic exploration of the aryl group allowing introduction of various substituents at the phenyl ring. The phenyl ring was replaced by heteroaromatics in another series of compounds.

EXPERIMENTAL

General

All purchased chemical reagents (Sigma-Aldrich, Brazil) were of the highest purity. Thin layer chromatography was done on aluminum sheets (DC Silicagel 60, Merck, Brazil) and observed under ultraviolet light at 254 and 366 nm in a Spectroline® ENF-260 CIF

chamber. NMR spectra (^1H : 400 MHz; ^{13}C : 100 MHz) were obtained from a Bruker® DPX 500 (Germany) device. All chemical shift values were recorded as δ (ppm), the coupling constant value J was measured in hertz; the peaks are presented as *s* (singlet), *d* (doublet), *t* (triplet), *dd* (double doublet), and *m* (multiplet).

Synthesis of target compounds

Guanylhydrazones **1-20** were prepared by the reaction of an appropriate aryl aldehyde with aminoguanidine hydrochloride (1:1.25 equiv). The reactants were dissolved in a minimal amount of methanol, heated to reflux, and stirred overnight. Upon cooling to room temperature, products crystallized out from the methanol solution. Products were recovered and dried under vacuum overnight. All reactions were monitored by TLC. The solvent was removed under reduced pressure and the solid was triturated with ethyl acetate yielding the substances as powders.

Assessment of cell viability

Cytotoxic effects of the synthesized compounds were evaluated against the following human cancer cell lines (National Cancer Institute, Bethesda, MD, USA): HCT-8 (colon carcinoma), SF-295 (glioblastoma) and MDA-MB-435 (melanoma). Tumor cells were seeded in 96-well plates (0.7×10^5 cells per well for adherent cells and 3×10^5 cells mL $^{-1}$ for suspended cells). After 24 hours, the compounds ($5 \mu\text{g mL}^{-1}$), dissolved in dimethylsulfoxide (DMSO), were added to each well using an automated workstation Biomek® 3000 (Beckman Coulter, USA) and incubated for 72 hours. Addition of doxorubicin (Sigma Aldrich Co., USA) at $5 \mu\text{g mL}^{-1}$ concentrations was performed in 150 μL of Roswell Park Memorial Institute (RPMI 1640) medium [2 mmol L $^{-1}$ of L-glutamine, 10 mmol L $^{-1}$ of 4-(2-hydroxyethyl)-1-piperazineethanesulfonic acid (HEPES), 1 mmol L $^{-1}$ of sodium pyruvate, 4.5 g L $^{-1}$ of glucose, 1500 mg L $^{-1}$ of sodium bicarbonate, 0.5 $\mu\text{g mL}^{-1}$ MTT]. After 3 hours, the formazan product was dissolved in 150 μL DMSO and the absorbance was measured using a multiplate reader (DTX 880 Multimode Detector, Beckman Coulter). The substance effect was quantified as the percentage of control absorbance at 595 nm.

Macrophage J774 cells were seeded (10^5 cells per well) in 96-well flat-bottom microplates. The cells were maintained for 2 hours at 37 °C, 5 % CO $_2$ and the medium was replaced with different concentrations of the drugs (10, 100 and 1000 $\mu\text{mol L}^{-1}$), in duplicate, and exposed for another 24 hours. Growth controls were also included. Afterwards, MTT solution was added to the cells and plates were returned to the incubator for another 4 hours to evaluate cell viability. This was followed by washing of the wells with 150 μL of DMSO for 15 minutes and measurement was performed spectrophotometrically at 540 nm.

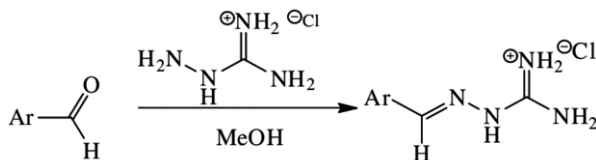
RESULTS AND DISCUSSION

Guanylhydrazone derivatives were obtained through the condensation reaction of benzaldehyde derivatives with aminoguanidine hydrochloride (**7**, **8**; see Scheme 1).

All compounds were characterized by hydrogen (^1H) and carbon (^{13}C) NMR. The main chemical shifts for each compound are shown in Table I. A signal at δ 7.27 to 9.05 ppm in ^1H NMR spectra of all synthesized substances could be observed and was attributed to

Table I. Spectral data of guanlylhydrazones

Compd.	¹ H NMR (δ, ppm, Hz)	¹³ CNMR (δ, ppm)
1	7.12 (t, 1H, <i>J</i> = 7.2), 7.19 (t, 1H, <i>J</i> = 7.2), 7.45 (d, 1H, <i>J</i> = 7.2), 7.8 (s, 1H), 8.28 (d, 1H, <i>J</i> = 7.2), 8.36 (s, 1H)	121.23, 122.75, 123.27, 124.31, 132.24, 137.51, 145.35, 155.20
2	7.45 (d, 2H, <i>J</i> = 8.5), 7.85 (d, 2H, <i>J</i> = 8.5), 8.15 (s, 1H)	129.23, 129.70, 132.81, 135.43, 146.01, 155.83
3	7.87 (d, 2H, <i>J</i> = 8.5), 8.04 (d, 2H, <i>J</i> = 8.5), 8.24 (s, 1H)	112.67, 119.11, 128.60, 133.02, 138.30, 145.36, 155.89
4	3.79 (s, 3H), 6.95 (d, 1H, <i>J</i> = 8.3), 7.18 (dd, 1H, <i>J</i> = 8.3 and 2.0), 7.33 (d, 1H, <i>J</i> = 2.0), 8.04 (s, 1H)	56.11, 112.05, 113.81, 121.09, 126.63, 147.06, 147.62, 150.49, 155.61
5	3.80 (s, 6H), 6.97 (d, 1H, <i>J</i> = 7.9), 7.23 (d, 1H, <i>J</i> = 7.9), 7.54 (s, 1H), 8.09 (s, 1H)	109.17, 111.60, 123.09, 126.53, 147.39, 149.51, 151.47, 155.63, 155.70, 156.16
6	3.81 (s, 3H), 6.83 (d, 1H, <i>J</i> = 8.0); 7.13 (dd, 1H, <i>J</i> = 8.0 and 1.8), 7.50 (s, 1H), 8.04 (s, 1H)	56.24, 110.08, 115.67, 123.18, 125.22, 147.74, 148.51, 149.78, 155.52
7	6.83 (d, 2H, <i>J</i> = 8.5), 7.66 (d, 2H, <i>J</i> = 8.5), 8.06 (s, 1H)	116.06, 124.79, 129.85, 147.52, 155.62, 160.28
8	7.63 (d, 2H, <i>J</i> = 8.5), 7.81 (d, 2H, <i>J</i> = 8.5), 8.17 (s, 1H)	124.31, 129.93, 132.15, 133.15, 146.11, 155.79
9	6.05 (s, 2H), 6.94 (d, 1H, <i>J</i> = 8.0), 7.17 (dd, 1H, <i>J</i> = 8.0 and 1.6), 7.64 (d, 1H, <i>J</i> = 1.6), 8.07 (s, 1H)	102.03, 105.94, 108.65, 124.80, 128.37, 146.83, 148.45, 149.80, 155.74
10	1.38 (s, 18H), 7.50 (s, 2H), 8.09 (s, 1H)	30.69, 35.01, 124.99, 139.66, 148.96, 155.53, 156.98
11	7.90 (s, 4H), 8.20 (s, 1H)	128.07, 130.01, 132.45, 137.86, 146.18, 155.84, 167.36
12	7.45 (m, 1H), 7.57 (d, 2H, <i>J</i> = 7.9), 8.42 (s, 1H)	129.56, 130.01, 132.18, 134.53, 142.99, 155.78
13	7.72 (t, 1H, <i>J</i> = 7.5), 7.84 (t, 1H, <i>J</i> = 7.5), 8.09 (d, 1H, <i>J</i> = 8.2), 8.40 (d, 1H, <i>J</i> = 8.2), 8.18 (d, 1H, <i>J</i> = 4.3), 8.98 (d, 1H, <i>J</i> = 4.3), 9.05 (s, 1H)	119.08, 123.89, 125.31, 128.12, 130.14, 130.29, 137.09, 143.47, 148.70, 150.61, 155.77
14	7.47 (m, 5H), 7.74 (d, 2H, <i>J</i> = 8.4), 7.94 (d, 2H, <i>J</i> = 8.4), 8.24 (s, 1H)	127.20, 127.35, 128.44, 128.68, 129.51, 132.94, 139.68, 142.41, 146.95, 155.79
15	7.43 (m, 3H), 7.84 (m, 2H), 8.19 (s, 1H)	128.03, 129.17, 130.99, 133.81, 147.35, 155.80
16	3.84 (s, 3H), 7.98 (s, 4H), 8.25 (s, 1H)	2.79, 128.19, 129.87, 131.21, 138.25, 146.08, 155.82, 166.30
17	7.68 (d, 1H, <i>J</i> = 8.4), 7.79 (dd, 1H, <i>J</i> = 8.4 and 2.0), 8.24 (d, 1H, <i>J</i> = 2.0), 8.16 (s, 1H)	128.45, 129.02, 131.33, 132.25, 133.13, 134.66, 144.75, 155.81
18	7.47 (m, 2H), 7.74 (d, 1H, <i>J</i> = 8.0), 8.06 (s, 1H), 8.17 (s, 1H)	126.84, 127.30, 130.58, 131.04, 134.19, 136.05, 145.79, 155.81
19	7.36 (m, 5H), 7.49 (m, 4H), 8.06 (s, 1H)	126.84, 127.30, 130.58, 131.04, 134.19, 136.05, 145.79, 155.81
20	7.60 (t, 2H, <i>J</i> = 7.8), 7.70 (d, 1H, <i>J</i> = 7.8), 8.12 (d, 1H, <i>J</i> = 7.8), 8.21 (s, 1H), 8.31 (s, 1H)	124.17, 125.83, 127.15, 130.29, 132.20, 134.95, 145.92, 123.12, 155.74



Scheme 1

H from imine, since this position is highly deshielded due to the induced anisotropy generated by the bound aromatic ring and the imine double bond. In ¹³C NMR spectra, the signal related to the imine carbon is usually seen around δ 145 ppm, whereas the signal ranging from δ 155 to 175 ppm is associated with the highly deshielded quaternary carbon of guanidine moiety.

The cell lines were exposed to derivatives guanylhydrazones (0–5 $\mu\text{mol L}^{-1}$) and positive control: doxorubicin (5 $\mu\text{mol L}^{-1}$) for 72 hours. Experiments were performed in triplicate.

Phenylaminoguanidine **15** exhibits a very low cytotoxic activity, only 20 % inhibition of tumor cell lines at 5 $\mu\text{mol L}^{-1}$. Introduction of an electron-donor group such as hydroxyl or methoxyl **4**, **5**, **6** and **7** at *ortho*- and/or *para*-position did not increase the cytotoxic activity. However, bridging between two groups in **9** induced a slight increase in activity towards the MDA-MB-435 cell line (Table II).

Replacement of the electron-donor group by electron-withdrawing polar groups such as nitrile **3** or formyl **11** showed no significant improvement in potency relative to the parent phenyl group. Introduction of halogen atoms significantly enhanced the activity, as seen in compounds **2**, **8** and **18**.

Compound **6** has shown higher selectivity against MDA-MB-435 than the other two cell lines. This fact, may be related to hydroxyl at position 3 in the phenyl ring, showing superior activity compared to the compound **4**.

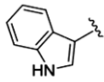
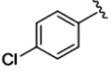
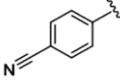
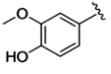
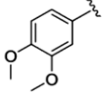
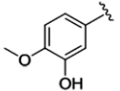
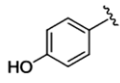
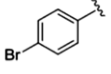
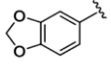
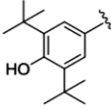
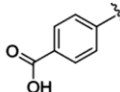
A significant increase in cytotoxic activity was observed, when chlorine was introduced in *para*- position, since compound **2** was found more potent than **18** (*meta*-position). However, disubstituted 3,4-dichloro analogue **17** led to better activity, with 100 % inhibition of tumor cell lines at 5 $\mu\text{mol L}^{-1}$. Interestingly, guanabenz (trade name, Wintensy®) **12**, a drug developed as an alpha-2 adrenergic agonist was significantly less active than compound **19**.


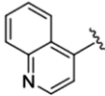
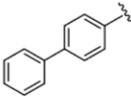

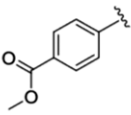
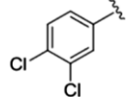

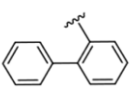

Introduction of a second phenyl ring at *para*- **14** or *ortho*- **19** position led to a set of more potent compounds. This effect highlights the importance of the lipophilic character, which probably plays an important role in cytotoxic activity. In this sense, compound **10** was also highly active, since the presence of the bulky *tert*-butyl groups increased lipophilic character to the molecule, resulting in the best compound of the series.

Finally, the replacement of the phenyl ring by a heteroaromatic ring as in **1** and **13** led to activity decrease compared to compound **15**.

The most powerful compounds recognized in cytotoxic assays against human cancer cell lines at 5 $\mu\text{mol L}^{-1}$ (Table II) were tested to obtain IC_{50} values, which were compared to those of doxorubicin (control), as shown in Table III. Each assay was carried out in dupli-

Table II. Cytotoxicity of guanylhydrazone derivatives against human cancer cell lines

Compd.	Inhibition (%) ^a			Ar	Yield (%)
	HCT-8	MDA-MB-435	SF-295		
1	–	–	24.0		71
2	72.9	80.2	72.1		71
3	25.4	38.8	28.6		89
4	3.5	0.2	25.3		93
5	–	17.3	19.5		98
6	–	32.4	2.5		92
7	–	–	11.5		95
8	67.3	81.6	58.0		83
9	–	53.2	19.6		81
10	97.2	92.6	97.8		85
11	10.3	16.3	26.2		71

12	45.2	15.3	51.9		89
13	42.1	29.8	52.0		91
14	100	100	100		90
15	28.3	26.5	24.7		89
16	43.1	48.6	37.8		92
17	100	100	100		92
18	52.1	46.0	52.0		91
19	96.9	95.2	93.8		92
20	52.7	33.7	45.5		88
Doxorubicin	97.3	96.9	87.6	–	–

^a Average value for 5 $\mu\text{mol L}^{-1}$ concentration ($n = 3$).

– Not active

cate and the means and respective confidence intervals were obtained by nonlinear regression analyses.

In order to investigate the effect of guanylylhydrazone derivatives on non-tumor cells, compounds **10**, **14**, **17** and **19** were evaluated on macrophages J774 after 24 hours of incuba-

Table III. IC_{50} values of guanylhyazone derivatives in three human cancer cell lines

Cell line	IC_{50} ($\mu\text{mol L}^{-1}$)				Doxorubicin
	10	14	17	19	
MDA-MB-435	3.4	4.2	0.4	9.8	0.8
HCT-8	3.9	1.8	0.7	3.6	0.02
SF-295	3.9	2.0	0.6	6.0	0.3

tion and data obtained are displayed in Fig. 2. Results suggest that **10** shows low toxicity to non-tumor cells, because it is only able to decrease cell viability at a high concentration of 1 mmol L^{-1} . The other three compounds, **14**, **17** and **19**, showed higher cytotoxicity for non-tumor cells since they reduced cell viability at all tested concentrations (10 , 100 and $1000 \mu\text{mol L}^{-1}$). This confirms that compound **10** acts selectively on tumor cell lines and bears the most promising profile regarding cytotoxic activity.

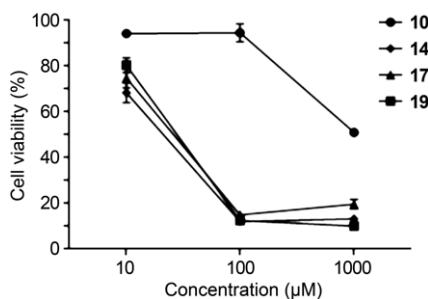


Fig. 2. J774 cell viability after 24 h in the presence of selected guanylhyazone compounds (mean \pm SD, $n = 2$).

CONCLUSIONS

The results reported in this study indicate that lipophilic guanylhyazones **10**, **14**, **17** and **19** showed the highest cytotoxic activity against different human cancer cell lines. Compound **10** showed higher selectivity towards tumor cell lines and therefore it may be considered promising for the development of novel cytotoxic agents based on this scaffold.

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Supplementary material is available from the authors upon request.

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